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(21) International Application Number: PCT/AU93/00267 (22) International Filing Date: 4 June 1993 (04.06.93) (30) Priority data: PL 2789 4 June 1992 (04.06.92) AU (71) Applicant (for all designated States except US): BETATENE LIMITED (AU/AU); 71-73 Taunton Drive, Cheltenham, VIC 3192 (AU) (72) Inventor; and (75) Inventor/Applicant (for US only): SCHLIPALIUS, Lance, Ellison (AU/AU); 71-73 Taunton Drive, Cheltenham, VIC 3192 (AU). (74) Agent: McMASTER, Wayne; Freehill Patent Services, Level 47, 101 Collins Street, Melbourne, VIC 3000 (AU).		(81) Designated States: AT, AU, BB, BG, BR, CA, CH, CZ, DE, DK, ES, FI, GB, HU, JP, KP, KR, KZ, LK, LU, MG, MN, MW, NL, NO, NZ, PL, PT, RO, RU, SD, SE, SK, UA, US, VN, European patent (AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG). Published With international search report.	
(54) Title: HIGH CIS BETA-CAROTENE COMPOSITION			
(57) Abstract A carotenoid composition derived from a natural source wherein at least 50 % by weight of the carotenoid content of the composition is cis beta-carotene and preferably 9 cis beta-carotene. Typically, the beta-carotene content of the composition is predominantly 9 cis beta-carotene.			

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1.

HIGH CIS BETA-CAROTENE COMPOSITION

FIELD OF THE INVENTION

The invention relates to a carotenoid composition derived from a natural source, with a high cis beta-carotene concentration and its preparation from natural sources, and more preferably a high/ 9\ cis beta-carotene composition and its preparation from natural sources.

BACKGROUND OF THE INVENTION

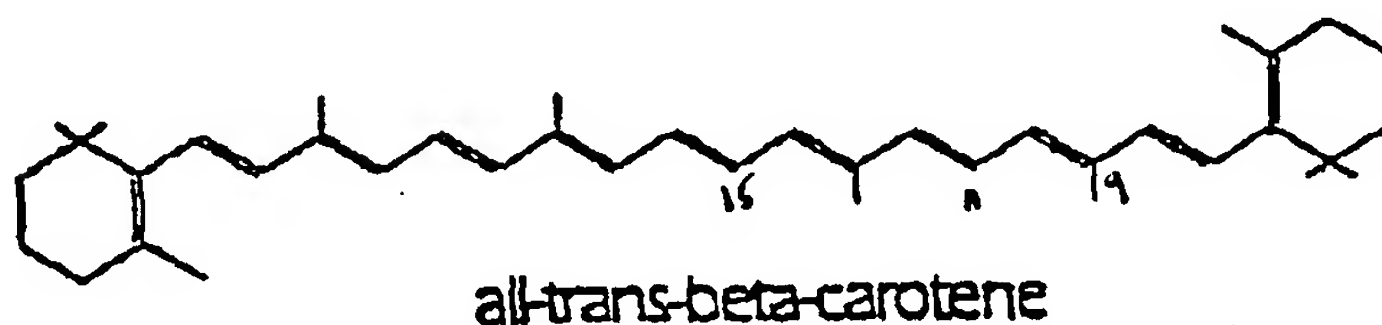
In this specification it is to be understood that the natural sources of carotene include fruits, vegetables and other plant tissue, and animal tissue. A particularly important commercial source of carotene is certain types of algae.

Beta-carotene occurs in a number of different chemical isomer forms. Some of these are geometrical isomers which have a different orientation around one of the double bonds in the conjugated double bond structure of the molecule.

This can occur in a number of positions along the conjugated backbone to make a range of different geometrical isomers. In some cases there can even be more than one double bond where change of orientation occurs.

2.

The most common geometrical isomer is the all trans isomer with a structure occurring as shown as follows where the main carbon chain of the molecule occurs in a trans (across) or straight configuration.



5 However, there are cis forms of beta-carotene which occur naturally, and can be produced by chemical synthesis, or formed by physical processes like heat on the all trans isomers, where the main carbon chain of the molecule takes a bend (cis) or sideways configuration. Naturally occurring cis forms of beta-carotene are not known to occur over a 10 weight percentage of approximately 30% to less than 50% of the total carotenoid content.

Associated with the different geometric isomers are different properties and possible functions and for this reason there are potential benefits in relatively concentrated forms of the cis isomers.

3.

In natural products such as fruit, vegetables, algae and other plant and animal material the carotenoids are stabilized as part of the cell structure in small micron or sub-micron sized particles in the cell organelles or even by association with other molecules which stabilize the isomeric forms produced by the biochemical pathways of the organism. However, in the preparation of concentrated forms of these materials for commercial products desired from the natural sources, the natural stabilising capacity of the cellular structure may be removed or reduced in the extraction and concentration of the carotenoids.

- 10 In addition, as the carotenoid products are concentrated to increase their beta-carotene concentrations and to remove the other cell material which is not desired in the product, there is a natural tendency for certain isomers to crystallize out.

Crystallisation is a problem in certain applications since the crystalline form may not be available for efficient use in the application because of its relative insolubility. Crystallisation occurs particularly with all trans beta-carotene and as a result it is not, for example, readily available for biological use.

- 20 Cis isomers, on the other hand are much less likely to crystallise and as a consequence are much more soluble than the trans isomers. For this reason, it is often more desirable to use beta-carotene containing compositions with higher concentrations of cis isomers for various applications. For example, the 9 cis isomer is much more readily soluble in oils than the all trans form. In fact, it is very difficult to get the

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9 cis isomer to crystallize out from naturally derived oils, thus making it difficult or expensive to purify on a large scale.

Unlike trans isomers, cis isomers have a number of applications, including use in water dispersible food colourants and in tabletting powders. The
5 cis isomers are also useful in a water dispersible form as emulsions for colouring and in beverages. They can also be used in special vitamin supplements in concentrates for direct supplementation or as part of food.

In naturally occurring products the proportion of cis isomers is rather small, but one of the highest proportions occurs in the halophilic alga
10 *Dunaliella salina* where normally 30% to below 50% of the total carotenoid content occurs as the 9 cis form.

The present invention is accordingly directed to compositions with a high cis beta-carotene composition derived from natural sources.

DESCRIPTION OF THE INVENTION

15 Accordingly, in one form of this invention, a carotenoid composition derived from a natural source is produced, wherein at least 50% by weight of the carotenoid content of the composition is cis beta-carotene and preferably 9 cis beta-carotene. Typically, the beta-carotene content is predominantly 9 cis beta-carotene.

20 In another form of the invention, the preferred range of cis beta-carotene is between at least 50% and 80%, and more preferably between 60% and 70%. Another preferred range is between 60% and 85%.

5.

In yet another preferred form of the invention, the high cis beta-carotene from natural sources is derived (for example, by concentration and purification by physical means) from a product of lower concentrations of the cis beta-carotene to a concentration of at least 70%
5 cis beta-carotene. Preferably this may be achieved by the removal of the substantially all trans beta-carotene using physical processes.

These high percentage cis isomer compositions have been found to exhibit high solubility and are readily available for physiologically active purposes. This is thought to be because of the following factors in
10 particular:

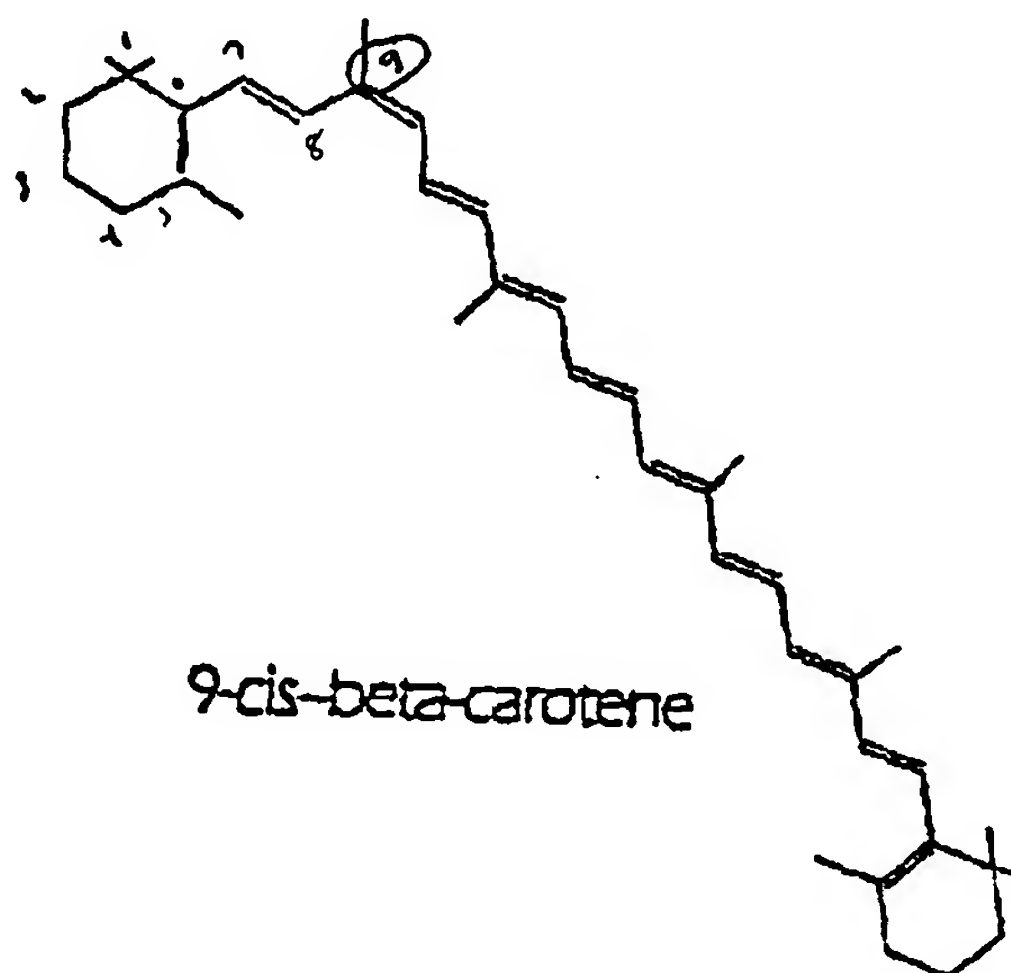
- (a) cis isomers occur in human body tissues in significant amounts and since the action of metabolism occurs in the tissues it is possible that the cis isomers have a physiological function;
- (b) cis isomers are typically lower in the bloodstream which is higher in
15 the all trans isomer; this suggests the cis isomers may be rapidly taken into the tissues; and
- (c) cis isomers appear to be easily absorbed from the intestine.

For example, high percentage cis isomers would be expected to have improved effect in use in medical applications, for example, in the healing
20 and prevention of cancer, cardiovascular disease and other illnesses, since the cis isomers, because of their solubility, are likely to preferentially accumulate in the tissues being removed from the bloodstream over the trans form.

6.

The improved solubility is also likely to assist in the application of the high percentage cis isomer in the topical application of the composition and for ease in the preparation in food colour applications. The cis isomers may assist in the stability of emulsions or powders of 5 beta-carotene for commercial purposes which is partly due to their solubility characteristics.

The structure of the 9 cis isomer of beta-carotene is as follows:



This 9 cis isomer of beta-carotene is preferably derived from particular 10 natural sources of plant products including green peppers, apricots, flowers of certain species of the Acacia genus, cucurbitaceae and in the alga, *Dunaliella salina*, which has the highest concentration of the 9 cis isomer of such sources. In this regard, see Ami Ben-Amotz, Amnon Lers and Mordhay Aron: "Steroisomers of Beta-Carotene and Phytoene in 15 the Alga *Dunaliella bardawil*" Plant Physiol. (1988) 86, 1286-1291

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(*Dunaliella bardawil* has subsequently been acknowledged by Ami Ben-Amotz as naturally occurring *Dunaliella salina*).

The proportion of the total cis isomers (predominantly the 9 cis isomer) content in the total carotenoid content of the alga *Dunaliella salina* is normally found at around 30% to below 50% of the total carotenoid content on a weight basis as determined by the high pressure liquid chromatography (see method below commencing at page 14) and visible light spectrophotometry techniques.

Preferably, the increase of the 9 cis isomers in the composition is to about 70% of the total carotenoid absorbance and more preferably 80% of the total carotenoid content. The remaining 20% typically would consist of all trans beta-carotene (for example 10%), other cis forms (for example 5%) and other carotenoids (for example 5%).

The finished product is preferably dispersed in a natural carrier oil from animal, vegetable and mineral origins and particularly olive, corn, soya bean, essential oils, terpene based oils and fish derived oils.

The provisions of small quantities of oil soluble anti-oxidants may be beneficial for certain applications of the product. Examples of these would be butylated hydroxy anisole, butylated hydroxytoluene, propyl gallate, ethoxyquin and ascorbyl palmitate plus other natural antioxidant extracts, for example, derived from herbs and preferably natural tocopherols.

8.

The anti-oxidants may be used to assist to protect the high cis beta-carotene preparations from oxidation, which is relatively more important when in a lower total beta-carotene concentration product, for example, less than 5% of beta-carotene in the preparation. However, 5 even at higher concentrations it is important to protect the beta-carotene preparations from oxidising.

A concentration of 0.01% to 1.0%, preferably 0.01% to 0.5%, of the pure active anti-oxidant is typically used, depending on the actual anti-oxidant of choice and application, as this concentration range has been shown by 10 experience to be sufficient for normal protection of the ultimate composition over its shelf life.

EXAMPLES OF METHODS TO PRODUCE THE HIGH CIS ISOMER COMPOSITION

The product is prepared by a series of physical unit operations 15 Examples of methods used to produce high cis isomer compositions, and more preferably to produce high 9 cis isomers are set out below as Example A and Example B.

Example A demonstrates a method using centrifugation (which relies on the principle of separation by density difference). The crystalline 20 fraction containing the trans crystal is separated from the cis isomer soluble oil fraction.

9.

This type of separation can either be done in a batch type centrifuge, usually on a small scale or for laboratory use or in a continuous machine to process larger volumes on an industrial scale.

The density of the all trans isomer crystals is greater than the cis isomers (including the 9 cis isomer) when in solution in vegetable oils having a relative density of about 0.92 grams per cubic centimeter. When the temperature or melting point of the carrier vegetable oil permits the oil to be liquid and of a viscosity to allow the all trans crystals to settle when the centrifugal force is applied, centrifugation will separate the all-trans isomer from the cis isomers. Being of the greater density, the all trans crystals can be separated by sedimentation and leave the machine in the heavy fraction and the cis isomers stay in the soluble vegetable oil or light fraction.

Centrifuges (for example, an Alfa Laval model number FUVPX207) which are capable of opening the bowl to remove a semi solid heavy sludge can concentrate the all trans crystals in this sludge thus increasing the yield of the high cis light fraction. The flow rate of material through the centrifuge has a large bearing on the result so this has to be optimized for the machine concerned.

Example B relies on filtration using a basket type centrifuge and filter bag, however, it will be readily understood by a person skilled in the art that the filtering operation can be performed on a range of filter materials, including a filter press, drum filter, filter beds, filter cartridge systems and more sophisticated membrane type filters. Any

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filter system can be used that employs pads, paper, cartridge filters or a centrifugal filtration using a filter in a basket centrifuge.

The crystals of all trans beta-carotene are held by the filter material but the soluble all cis material can pass through the filter thus increasing the proportion of cis isomers in the filtrate, that is, the product passing through the filter material.

Example A.

Ten grams of 4% natural beta-carotene in soya bean oil (commercially available from Betatene Limited and sold under the brand name "Betatene Limited 4% Natural Beta-Carotene in Soya Bean Oil") was placed in an 18mm diameter glass centrifuge tube and centrifuged at 2000 revolutions per minute for 15 minutes at 21°C in a Clements 2000 laboratory centrifuge.

At the conclusion, the supernatant oil (the light fraction) was recovered by removal with a Pasteur pipette and assayed for cis and trans beta-carotene content. After draining the residual supernatant from the heavy fraction, the same analysis was performed on the pelleted heavy fraction.

The percentages of the cis (including the 9 cis) beta-carotene and all trans beta-carotene (determined using the high pressure liquid

11.

chromatography method commencing on page 14) were as follows:

	CIS total predominantly 9 cis % of total carotenoid by weight	ALL TRANS
5 Original 4%	42	45
Light fraction	62	24
Heavy fraction	19	74

with the crystals being collected in the heavy phase or the sludge
fraction and the high cis product as the light fraction. The residue
10 includes other carotenoids.

The same procedure can be performed on a continuous production scale
using an open bowl desludging centrifuge such as an Alfa Laval model
number FUVPX 207. The flow rate of the feed to the centrifuge, the
setting of the skimmer to recover the high cis light fraction and the
15 desludging timing would need to be optimised for the machine or other
similar machines. This will be well understood by persons skilled in the
art.

Example B.

A crystalline suspension of beta-carotene in soya bean oil containing 21%
20 total carotenoid was filtered to retain crystals and to allow the soluble

12.

fraction to pass through. The all trans crystals can be separated from an oil based matrix, providing the oil is not too viscous to restrict movement of the oil through the filter bed. The starting material is available from Betatene Limited and is sold under the brand name

5 "Betatene Limited 20% Natural Beta-Carotene in Soya Bean Oil".

This operation can be achieved in a basket type centrifuge of a type like a Broadbent, Tolhurst for Burton with a 900 mm diameter bowl, or a similar machine. The bowl is made of perforated stainless steel plate and a fabric bag is placed in the bowl with a weave having gaps between the

10 fibres of approximately 20 to 25 microns in width. The 20% natural beta-carotene in soya bean oil is fed into the bag and the high trans crystals are retained on the bag letting the soluble high cis fraction pass through the bag and the perforations in the bowl. In the initial stages the filtrate may still contain the smaller crystals but this is overcome by

15 recirculation through the centrifuge. Before long the crystal bed on the inner surface of the bag acts as the main filter which will remove the small crystals. The operation continues till the flow of filtrate through the bag is too slow. The filtrate is collected as are the retained crystals at the end of the run and analysed for beta-carotene and cis-trans

20 isomeric profile.

The percentages of the cis beta-carotene and all trans beta-carotene (determined using the using the the high pressure liquid

chromatography method commencing on page 14) were as follows:-

	CIS	TRANS
	predominantly 9 cis	
	% of total carotenoid by weight	
5 Initial 21% oil suspension	32	63
Filtrate	82	14
Filtered crystals	27	68

The residue includes other carotenoids.

Example A is an actual example of a method that has been carried out on
10 a laboratory scale. Example B was carried out on a production scale.

It would also be understood by persons skilled in the art that cis isomers could be separated on a larger scale using preparative high pressure liquid chromatography. For example, Dunaliella salina algae which has a 9 cis isomer content of below 50% could be treated by preparative high
15 pressure liquid chromatography techniques to separate the isomers and obtain a very high purity product of the 9 cis isomer (for example, 80% or more of the 9 cis isomer). This separation technique relies on the variable retention of different chemical materials to a solid phase when the materials in various mixtures of volatile organic solvents are pumped
20 through the solid phase, usually in the form of spheres in a column. By collecting fractions at the end of the column the chemical materials

14.

may be separated from each other. The volatile organic solvents are evaporated from the pure chemical to provide the solvent free product.

EXAMPLE OF ISOMERIC ANALYSIS OF A CAROTENOID COMPOSITION
USING HIGH PRESSURE LIQUID CHROMATOGRAPHY

- 5 The following is an example of a standard method for conducting an isomeric analysis of a carotenoid composition using high pressure liquid chromatography to determine the percentage content of the cis isomers of beta-carotene and all trans beta-carotene. This was the method used in analysing the isomeric percentages in Example A and Example B above.
- 10 In summary, a sample of a beta-carotene containing composition in oil is dissolved in cyclohexane and diluted to a suitable concentration. Before injecting the resultant solution into the high pressure liquid chromatograph, the sample should be diluted with a mobile phase. The concentration of beta-carotene is determined by obtaining an absorbance
- 15 at a specific wavelength using a known extinction coefficient. The cis and trans contents are determined by separating the isomers by high pressure liquid chromatography. The percentages of cis isomers and trans isomers are then determined as a percentage of the total carotenoid content.

20 (1) Reagents and Equipment

The following is a list of the reagents and equipment that can be used in the analysis.

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Spectrophotometer with 10mm glass cells

Analytical balance

Cyclohexane AR grade

100 ml volumetric flasks

5 50 ml volumetric flasks

2.0 ml bulb pipettes

Chloroform AR grade

High pressure liquid chromatograph incorporating:

Isocratic pump

10 Injector able to handle up to 50µl

UV/VIS detector set at 453nm

Vydac 201TP54 reverse phase column 250mm

Acetonitrile HPLC grade

Methanol HPLC grade

15 10ml volumetric flask

(II) Preparation of the mobile phase

Prior to the analysis being conducted, a mobile phase should be prepared by weighing 400g of acetonitrile into a 1000ml storage bottle, adding 300g methanol and mixing well. The resultant mixture should be
20 adjusted to room temperature before being used in the analysis.

(III) Procedure

The method used in the actual analysis is set out below.

16.

A sample of beta-carotene in oil is weighed accurately (to within 0.01g) to the equivalent to 80mg of beta-carotene into a 100ml volumetric flask. Approximately 5ml of chloroform is added and mixed well until the sample has dissolved. To ensure that the sample has completely dissolved it should be viewed against a light source. If the sample is not completely dissolved, it should be allowed to stand for approximately 5 minutes. It is also possible to add a further 5ml of chloroform and to warm the mixture under, for example, hot tap water.

The volume of the mixture is then diluted with cyclohexane and mixed well. This is solution A. 2ml of solution A is then pipetted into a 50ml volumetric flask and diluted to volume with cyclohexane and mixed well. This is solution B. 2ml of solution B is then pipetted into a 50ml volumetric flask and diluted to volume with cyclohexane and mixed well. This is solution C. The absorbance of solution C is then measured at 455nm against a cyclohexane blank (this gives the total beta-carotene content and not the isomeric ratio). 1ml of solution B is then pipetted into a 10ml volumetric flask, diluted to volume with the carrier solvent and mixed well. This is solution D.

(iv) Standards

For comparative purposes, a standard is then prepared. The standard is prepared to provide a reference material of known concentration in the assay and to determine reference retention time ("R.T.") for the all trans isomer. The standard is made up according to the same procedure used to make up the sample. Weigh the reference standard material (Sigma all trans beta-carotene from Sigma Corporate in St Louis USA) to

the equivalent of 80mg of beta-carotene, then proceed as for the sample. The concentration of the standard is determined in the same way as the sample.

(v) Determination

5 The standard and the sample as prepared above are ready for injection.

The flow rate of the high pressure liquid chromatograph should be set to 1.0ml/minute. 20 µl of the standard is injected into the high pressure liquid chromatograph and allowed to run for 25 minutes. The injection concentration and volume are recorded. 20µl of solution D is then injected
10 into the high pressure liquid chromatograph and allowed to run for 25 minutes. The injection concentration and volume are recorded.

An example of a typical chromatogram is reproduced on page 18.

(vi) Calculation

I - Calculation for Determination of Beta-carotene Concentration

15 The bracketed figures are typical figures.

Absorbance of solution C at 455nm	Abs	(-0.4)
Weight of sample	m	(-0.5g)
Extinction coefficient E ₁	2500	cm ⁻¹ g ⁻¹
Injected concentration (mg/l)	I	(-4)

20

$$I = \text{Abs} \times 25 \times 10,000 / (2,500 \times 10) = \text{Abs} \times 10$$

II - Calculation for Determination of Cis and Trans Percentages

	Peak area of all-trans beta-carotene	T	(17%) R.T.- 13.5 minutes
5	Peak area of unknown cis isomer 1	C1	R.T.- 14 minutes (not seen in above chromatogram)
	Peak area of unknown cis isomer 2	C2	(9.8%) R.T.- 14.7 minutes
	Peak area of 9-cis isomer	C3	(43%) R.T.- 15.5 minutes
10	Peak area of 15-cis isomer	C4	(8%) R.T.- 16.68 minutes

The retention times used in the above table are the retention times that were seen in the above chromatogram. In the above chromatogram a peak was not seen for unknown cis isomer 1 ("the C1 peak"). Typically, this isomer will be seen after a retention time of about 14 minutes. The percentages given in the above table are approximate.

The individual peak areas are determined as a percentage by dividing the individual peak areas by the total peak area and multiplying the result by 100 to give the percentage of total carotenoid. These percentages appear under the heading "Conc" in the table that appears below the example chromatogram on page 18.

Alternatively, the following formula can also be used in calculating the cis isomers percentage of the total carotenoid content. The formula presumes

that a C1 peak is seen in the chromatogram:

$$\text{Cis \%} = (\text{C1} + \text{C2} + \text{C3} + \text{C4} / (\text{Total peak areas})) \times 100\%$$

Cis isomers are not always fully resolved but this does not effect their total cis isomer absorbance. In this regard, the C1 peak is not always
5 fully resolved or separated from the other cis isomers. In this case, the C1 peak can be neglected from the calculation.

The all trans isomer percentage of the carotenoid content can also be determined by the following formula:

$$\text{Trans \%} = (\text{T} / (\text{Total peak areas})) \times 100\%$$

10 When conducting this type of analysis it is recommended that low actinic glasswater is used as beta-carotene is degraded slowly in light. The analysis should also be performed in duplicate because the retention times are very dependant on temperature, the column and solvents should be maintained at 20°C.

15 Accordingly, the invention provides a novel carotenoid composition derived from a natural source, of high cis beta-carotene content.

21.

The claims defining the invention are as follows:

1. A carotenoid composition derived from a natural source, wherein at least 50% by weight of the carotenoid content of the composition is *cis* beta-carotene.
- 5 2. A carotenoid composition according to claim 1, wherein the *cis* beta-carotene content of the composition is between at least 50% and 80%.
3. A carotenoid composition according to claim 1, wherein the *cis* beta-carotene content of the composition is between 60% and 70%.
- 10 4. A carotenoid composition according to claim 1, wherein the *cis* beta-carotene content of the composition is between 50% and 85%.
5. A carotenoid composition according to any one of claims 1 to 4, wherein the composition is predominantly 9 *cis* beta-carotene.
- 15 6. A carotenoid composition according to any one of claims 1 to 5 wherein the composition comprises about 70% 9 *cis* beta-carotene.
7. A carotenoid composition according to claim 5 or claim 6 wherein the source of the 9 *cis* beta-carotene is selected from plant products including apricots, flowers of certain species of the *Acacia* genus, cucurbitaceae and *Dunaliella salina* and mixtures thereof.
- 20

22.

8. A carotenoid composition according to any one of claims 5 to 7,
wherein the source of the 9 cis beta-carotene is *Dunaliella salina*.
9. A carotenoid composition according to any one of claims 1 to 8,
wherein the composition is dispersed in a natural carrier oil
5 selected from animal, vegetable and/or mineral origins.
10. A carotenoid composition according to claim 9, wherein the oil
carrier is selected from olive, corn, soya bean, essential oils,
terpene based oils, fish derived oils and mixtures thereof.
11. A carotenoid composition according to any one of claims 1 to 10,
10 wherein the composition further comprises small quantities of oil
soluble anti-oxidants in the range of 0.01% to 1.0% by weight of
pure anti-oxidant.
12. A carotenoid composition according to any one of claims 1 to 10,
15 wherein the composition further comprises small quantities of oil
soluble anti-oxidants in the range of 0.01% to 0.5% by weight of
pure anti-oxidant.
13. A carotenoid composition according to claim 11 or claim 12, wherein
the anti-oxidant is selected from butylated hydroxy anisole,
butylated hydroxytoluene, propyl gallate, ethoxyquin and
20 ascorbyl palmitate plus other natural antioxidant extracts which
may be derived from herbs and preferably natural tocopherols,
and mixtures thereof.

23.

14. A method for producing a carotenoid composition according to any one of claims 1 to 13.

INTERNATIONAL SEARCH REPORT

international application no.

PCT/AU 93/00267

A. CLASSIFICATION OF SUBJECT MATTER Int. Cl. ⁵ C07C 403/24, A23L 1/30, 1/302, 1/303 According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) IPC A23L 1/30, 1/302, 1/303, C07C 403/24 Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched AU : IPC as above Electronic data base consulted during the international search (name of data base, and where practicable, search terms used) DERWENT : A23IC and C07C IC + CIS(S) CAROTENE		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to Claim No.
A	US,A, 3367985 (SURMATS) 6 February 1968 (06.02.68)	
A	US,A, 3441623 (SURMATS) 29 April 1969 (29.04.69)	
A	US,A, 3929757 (HOFFMAN-LA ROCHE AG) 29 April 1974 (29.04.74)	
<input type="checkbox"/> Further documents are listed in the continuation of Box C. <input checked="" type="checkbox"/> See patent family annex.		
* Special categories of cited documents : "A" document defining the general state of the art which is not considered to be of particular relevance "G" earlier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "Z" document member of the same patent family		
Date of the actual completion of the international search 27 July 1993 (27.07.93)		Date of mailing of the international search report 6 AUG 1993 (6.08.93)
Name and mailing address of the ISA/AU AUSTRALIAN INDUSTRIAL PROPERTY ORGANISATION PO BOX 200 WODEN ACT 2606 AUSTRALIA Facsimile No. 06 2853929		Authorized officer G. MASTERS <i>Gordon Masters</i> Telephone No. (06) 2432287

INTERNATIONAL SEARCH REPORT
Information on patent family members

International application No.
PCT/AU 93/00267

This Annex lists the known "A" publication level patent family members relating to the patent documents cited in the above-mentioned international search report. The Australian Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

Patent Document Cited in Search Report		Patent Family Member					
US	3441623	CH	503721	CH	511825	DE	1668215
US	3367985	CH	495349	DE	1618502		
US	3989757	AT	695674	BE	819259	CH	597112
		DE	2440747	FR	2242381	GB	1427690
		IT	1027535	JP	50050338	NL	7411403
		US	465202				
END OF ANNEX							

